

Table 26. Calculated compositions of liquid fractionates and crystalline residua derived from the high-alumina quartz tholeiite at 9–10 kb under wet conditions<sup>a</sup>

Pressure	9 kb	10 kb	10 kb	
Temperature	1,040° C	960° C	920° C	
Nature and estimated % of crystals extracted	Initial liquid	18% cpx 5% amph 2% opx 1% ilm	18% cpx 25% amph 2% opx 1% ilm	18% cpx 32% amph 3% plag 2% opx 1% ilm
Mol. prop.				
$\frac{100 \text{ MgO}}{\text{MgO} + \text{FeO}_{\text{Total}}}$	60.4	50.9	37.5	45.3
<i>Liquid fractionate</i>				
SiO <sub>2</sub>	52.9	55.9	59.7	64.5
TiO <sub>2</sub>	1.5	0.8	0.4	0.1
Al <sub>2</sub> O <sub>3</sub>	16.9	19.4	20.2	19.9
Fe <sub>2</sub> O <sub>3</sub>	0.3	0.4	0.5	0.7
FeO	7.9	7.2	5.8	3.7
MnO	0.2	0.3	0.3	0.4
MgO	7.0	4.4	2.1	2.0
CaO	10.0	7.8	6.7	3.9
Na <sub>2</sub> O	2.7	3.3	3.4	3.6
K <sub>2</sub> O	0.6	0.8	0.9	1.0
	100.0	100.3	100.0	99.8
<i>CIPW norm</i>				
Qz	5.5	14.4	25.3	
Or	4.8	5.4	5.9	
Ab	27.9	28.8	30.5	
An	35.7	33.2	19.3	
Cor	—	1.4	5.8	
Diop	2.4	—	—	
Hyp	21.9	15.3	12.1	
Mt	0.6	0.7	1.0	
Ilm	1.5	0.8	0.2	
<i>Crystal residuum</i>				
SiO <sub>2</sub>	44.8	43.0	42.7	
TiO <sub>2</sub>	2.5	2.7	2.7	
Al <sub>2</sub> O <sub>3</sub>	9.7	12.3	14.2	
FeO	9.1	10.1	11.2	
MgO	14.6	12.9	11.0	
CaO	16.6	13.8	14.8	
Na <sub>2</sub> O	1.0	1.8	1.9	
K <sub>2</sub> O	0.04	0.2	0.2	
	98.34	96.8	98.7	
<i>Mol. prop.</i>				
$\frac{100 \text{ MgO}}{\text{MgO} + \text{FeO}}$	74.2	69.5	63.7	

Table 27. Calculated compositions of liquid fractionates and crystalline residua derived from the dacite composition under wet conditions at 27 kb

Temperature		1,140° C	1,100° C
Nature and estimated % of crystals	Initial liquid	1% ga	9% ga 6% cpx
<i>Liquid fractionate</i>			
SiO <sub>2</sub>	65.0	65.3 <sup>a</sup>	68.8 <sup>a</sup>
TiO <sub>2</sub>	0.7	0.7	0.6
Al <sub>2</sub> O <sub>3</sub>	16.1	16.0	15.6
Fe <sub>2</sub> O <sub>3</sub>	1.4	1.4	1.6
FeO	3.5	3.3	1.5
MnO	0.1	0.1	0.1
MgO	1.8	1.7	0.6
CaO	5.0	5.0	3.8
Na <sub>2</sub> O	3.6	3.6	4.1
K <sub>2</sub> O	2.1	2.1	2.5
Mol. prop.	99.3	99.2	99.2
$\frac{100 \text{ MgO}}{\text{MgO} + \text{FeO}_{\text{Total}}}$	40.3	39.9	26.7
<i>CIPW norm</i>			
Qz	21.7	22.0	26.5
Or	13.0	12.4	14.8
Ab	30.5	30.5	34.6
An	21.5	21.3	16.8
Diop	2.3	2.9	1.7
Hyp	7.7	6.8	1.3
Mt	2.0	2.0	2.3
Ilm	1.3	1.3	1.1
<i>Crystal residuum</i>			
SiO <sub>2</sub>	39.4	43.2	
TiO <sub>2</sub>	1.5	1.3	
Al <sub>2</sub> O <sub>3</sub>	21.4	18.8	
FeO	19.8	15.1	
MnO	0.3	0.2	
MgO	8.9	8.4	
CaO	8.8	12.1	
Na <sub>2</sub> O	—	1.0	
K <sub>2</sub> O	—	—	
Mol prop.	100.1	100.1	
$\frac{100 \text{ MgO}}{\text{Mg} + \text{FeO}}$	44.5	49.8	

<sup>a</sup> Denotes compositions determined from analyses calculated in the manner described on p. 114.

garnet will constitute the residuum when dacitic or rhyodacitic liquids are extracted.

The alumina content of the liquids is high compared with the content normally found in members of the calc-alkaline series. This is attributed to the fact that the experiments have been conducted on a high-alumina basaltic composition. If a basalt with an alumina content of about 14% had been used then the alu-